## **HOMEWORK**

due March 14<sup>th</sup>

1.	A gold bar at Fort Knox weighs approximately 12.441 kg. How many atoms of gold are there in a Fort Knox gold bar? (Atomic weight of Au is 196.966)
2.	How many moles of sodium carbonate are present in 1.773 x $10^{17}$ atoms of sodium carbonate, $Na_2CO_3$ ?
3.	Nitroglycerin has the chemical formula $C_3H_5N_3O_9$ . How many moles of carbon are in there in 2.00 moles of nitroglycerin?
4.	How many carbon atoms are there in 52.06 g of carbon dioxide, $CO_2$ ? (Atomic weights C: 12.0107 O: 15.9994)